

Infrared spectroscopy of III-V and II-VI core-shell nanoparticles

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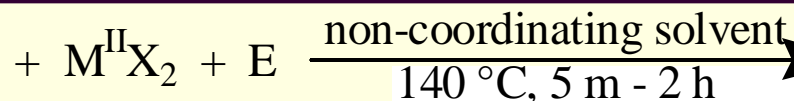
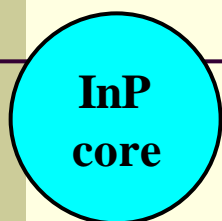
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Motivation and Objectives

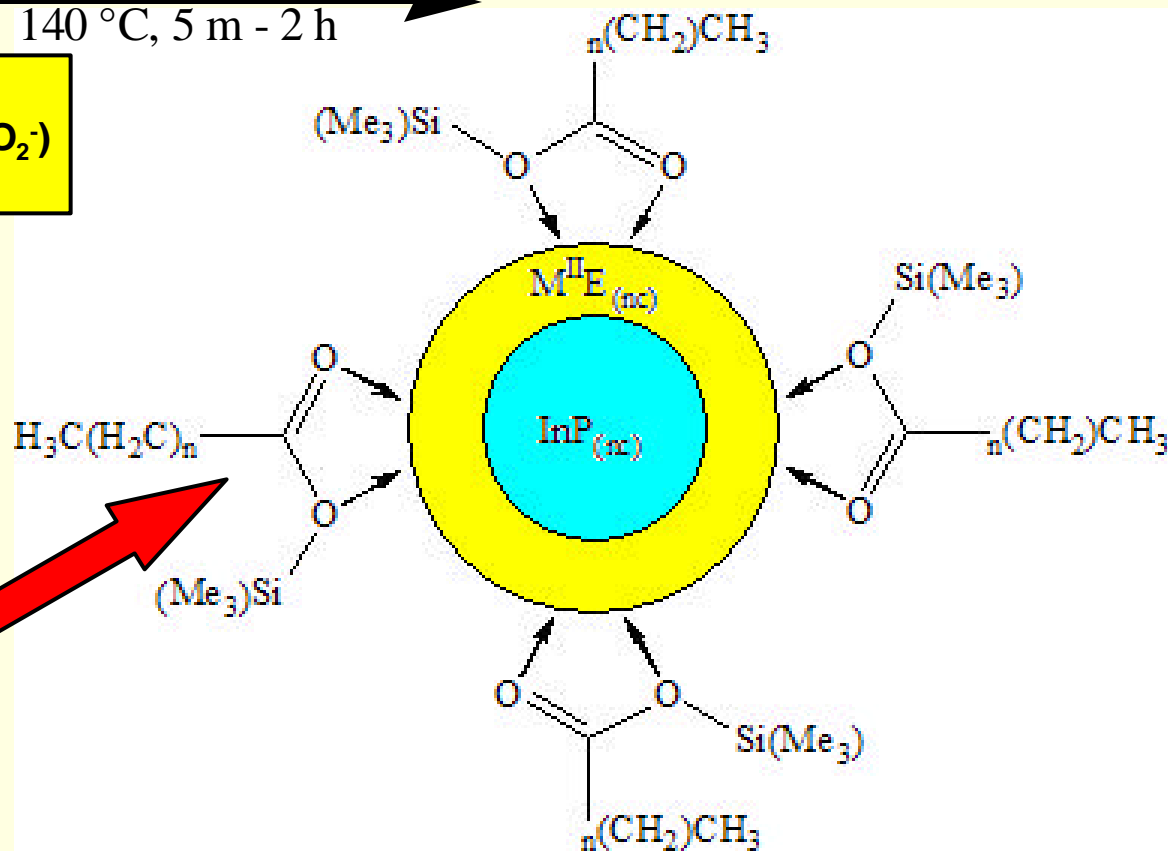
- Need for good nanoparticle light emitters in visible/near IR for communications and bio-medical applications \mathcal{P} **structure** and **dynamics** both important
- Recent progress in chemical fabrication of nanoparticles and core-shell nanoparticles with excellent emission properties
- **Vibronic properties important** \mathcal{P} both **structure** and **dynamics (non-radiative recomb.)**
 - Optical phonons in InP nanoparticles and InP/II-VI core-shell nanoparticles -- IR and Raman spectroscopies
 - Dependence of vibronic properties on nanostructure dimensions (surface/interface) optical phonon modes

Sample Preparation



$M^{II} = \text{Cd or Zn}$
 $X = \text{Acetate } (\text{CH}_3\text{CO}_2^-)$
 $E = \text{S or Se}$

- Generation of core InP nanocrystals with *In-situ* surfactant
- Non-coordinating solvent– octadecene
- Surfactant formed “in-situ” (none added)
- Different precursor ratios and temperature sequences explored



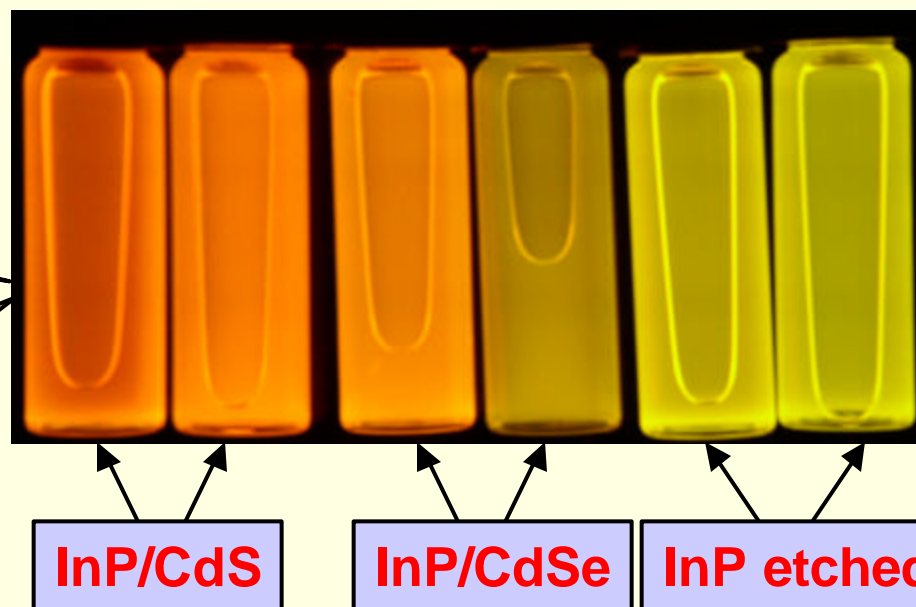
InP/II-VI
core-shell nanocrystal

Advantages of present approach (organic precursor)

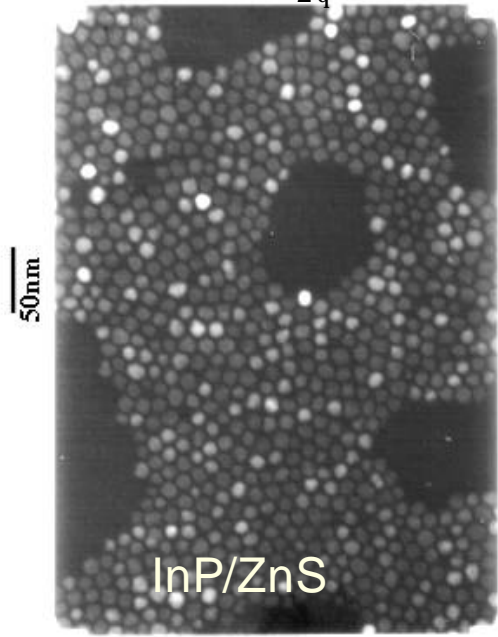
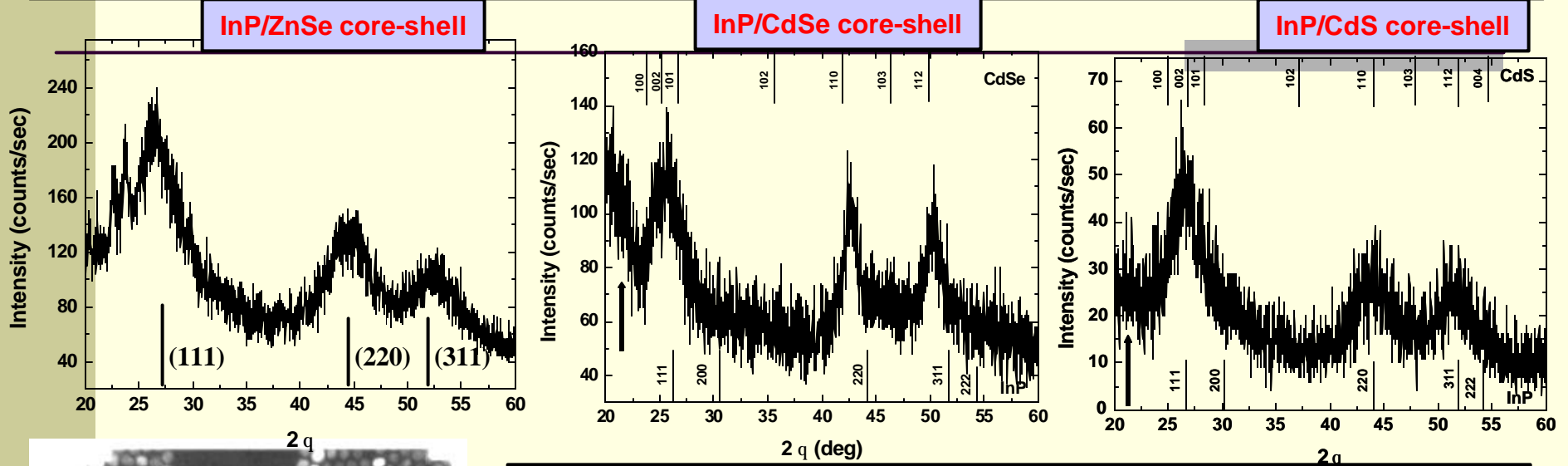
- Rapid fabrication and scalability of process
- Greater controllability (sizes and shapes of nanoparticles)
- Possibility of embedding nanoparticles in a wide range of matrices (liquids or solids)

**InP/II-VI nanocrystals
in ambient light**

**Shell thickness is
tunable by varying
amounts of shell
precursors**



Characterization of samples by TEM and XRD



- X-ray diffraction

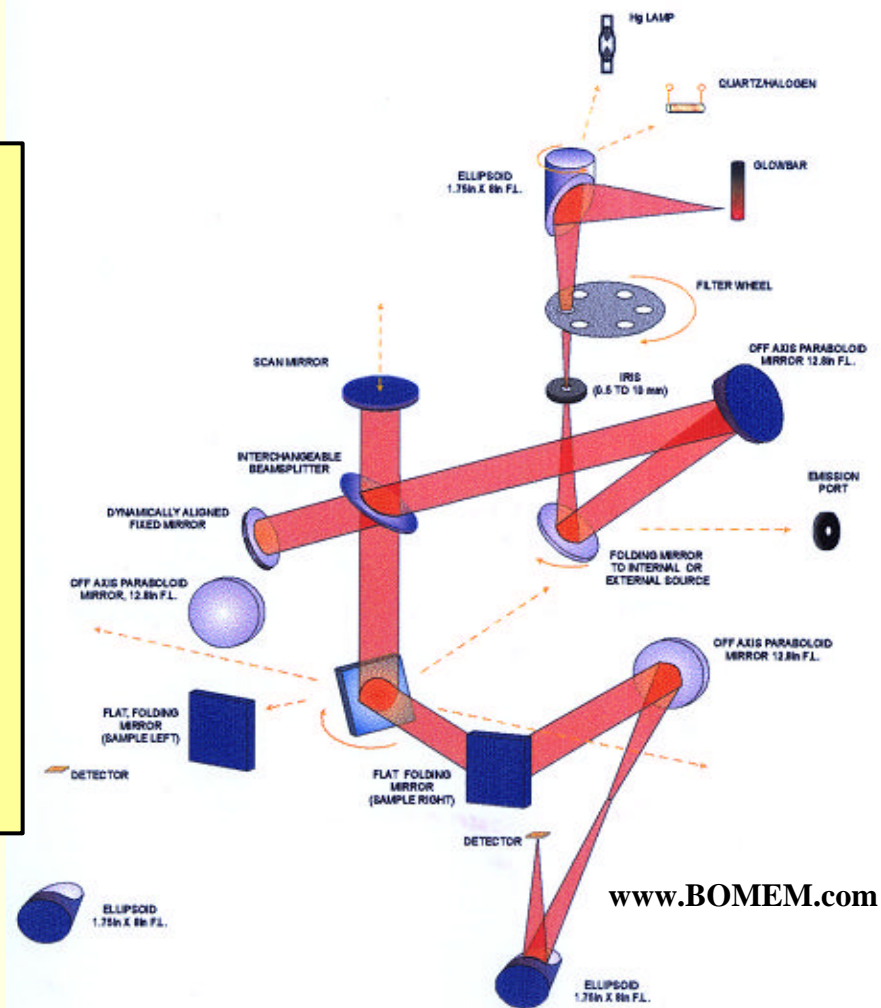
- ↳ peaks for (111), (220) and (311)
- ↳ CdS and CdSe samples appear to be wurtzite – ZnSe is zinc blende structure

- Transmission Electron Microscopy

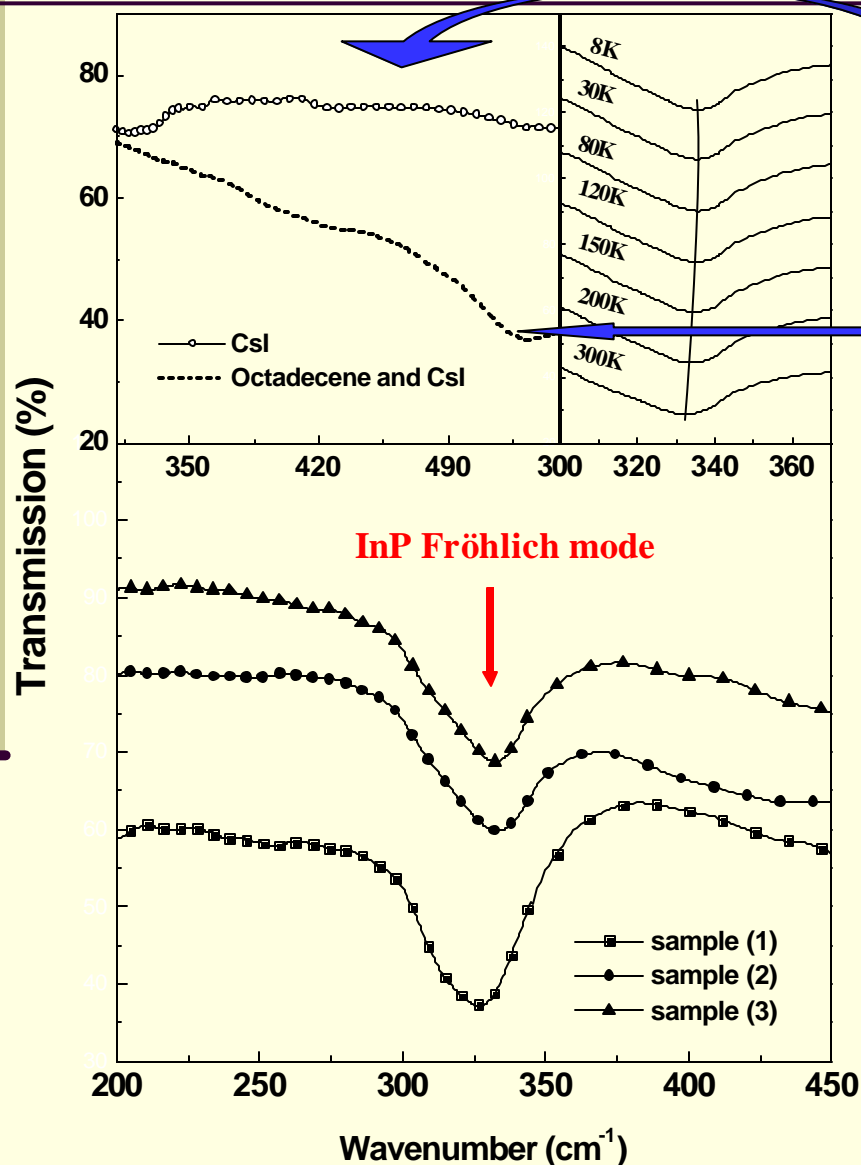
- ↳ Nanoparticles are crystalline
- ↳ Sizes between 5 and 20 nm

Samples and IR Spectroscopy

- **Samples for IR Measurements**
 - Nanoparticles in CsI matrix
 - ▷ pellets
- **Far IR transmission**
 - BOMEM Fourier Transform IR Spectrometer
 - ▷ Si bolometer detector
 - ▷ 6 μm Mylar beamsplitter (spectral range 20 - 370 cm^{-1})



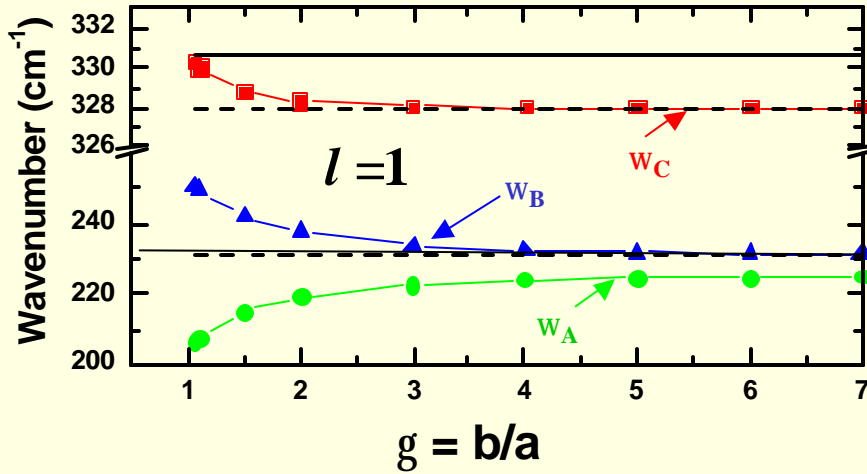
Results (InP nanoparticles)



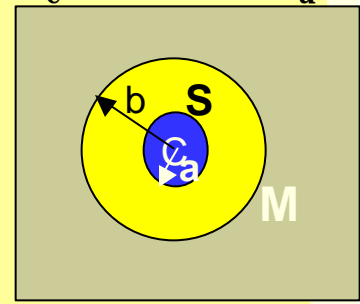
- CsI transparent in region of interest
- Solvent (octadecene) transmission spectrum does not interfere with characteristic bands of InP
- **Fröhlich mode**
 - Common feature in transmittance spectra of InP nanoparticles embedded in CsI matrix
 - Same temperature dependence as TO mode

InP/ZnSe core-shell

SO phonon modes from continuum dielectric model

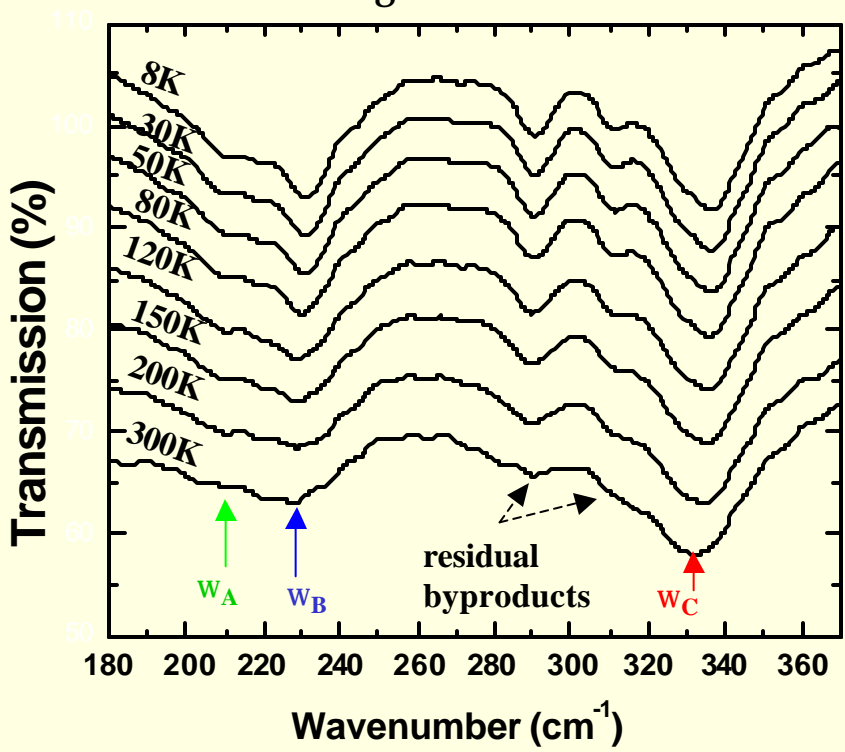


$$\frac{e^S(\omega)}{e^M(\omega)} = - \frac{e^C(\omega) \left[g^{2l+1} - 1 \right] + e^S(\omega) \left[\frac{l+1}{l} g^{2l+1} - 1 \right]}{e^S(\omega) \left[g^{2l+1} - 1 \right] + e^C(\omega) \left[\frac{l}{l+1} g^{2l+1} - 1 \right]}$$



γ = ratio of radius of shell (b) to core (a)

l = angular mom./spherical harmonic Quantum number

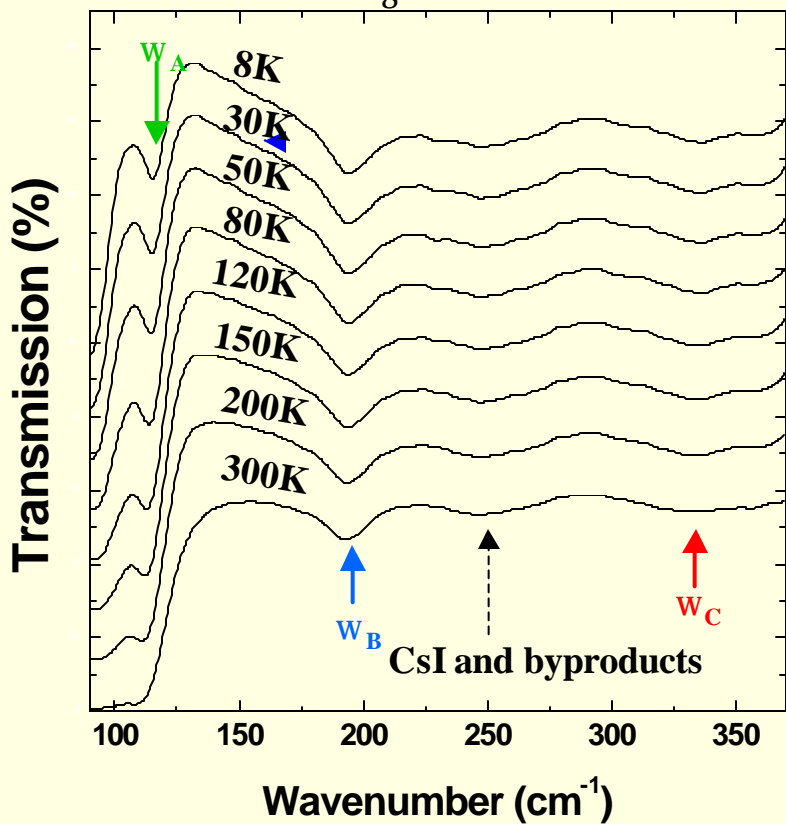
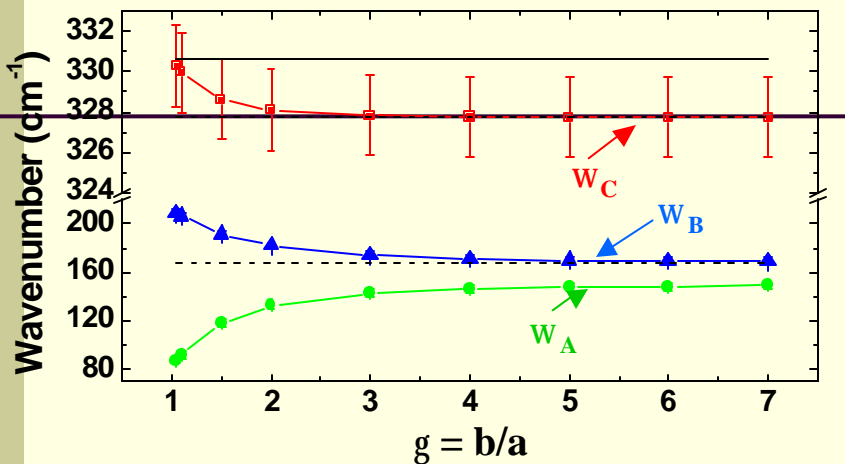


ω_C = core (InP) vibrational mode frequency (Fröhlich mode)

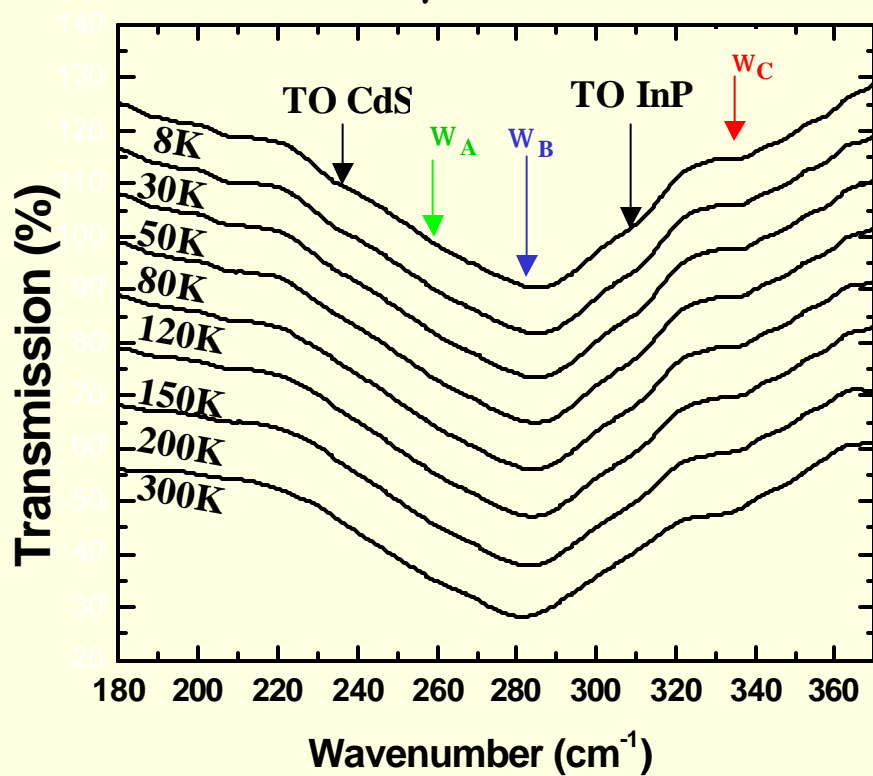
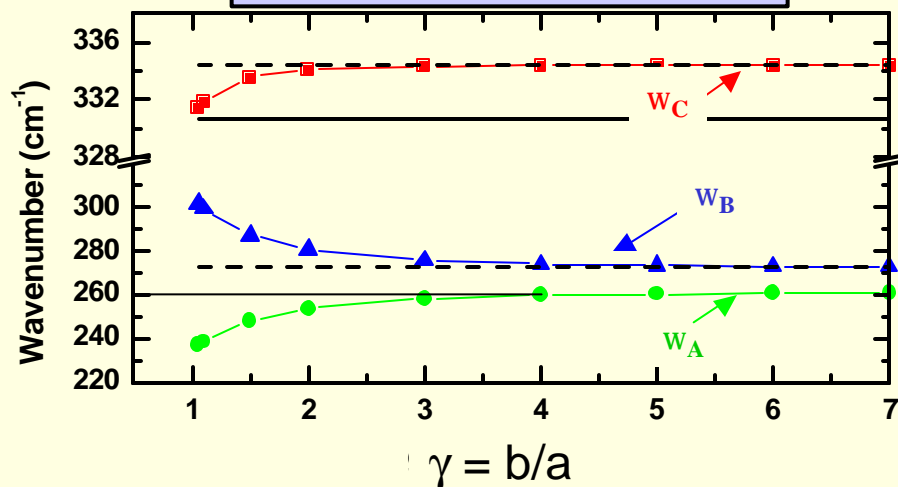
ω_B = surface vibrational mode frequency at interface between shell and matrix

ω_A = vibrational mode frequency of shell

InP/CdSe core-shell w/zinc blende phonons



InP/CdS core-shell w/zinc blende phonons



Summary and Conclusions

- **Comparison between theory and experiment**
 - InP/ZnSe $g \sim 1.4$ to 2.5
 - InP/CdSe $g \sim 1.2$ to 1.6
 - InP/CdS $g \sim 1.5$ to 3
- **For thin shell (eg. InP/ZnSe, InP/CdSe) three vibrational modes, ω_C , ω_B , and ω_A , are resolved**

- **InP nanoparticles**

- IR transmission spectra of InP nanoparticles exhibit common feature \Rightarrow **Fröhlich mode**

- **InP/II-VI core-shell nanoparticles**

- Good agreement between theoretical and IR experimental results
- IR spectroscopy can provide detailed characterization of their structure

**IR absorption spectroscopy good probe
of quality of core-shell nanoparticles – shell thickness**